

Course Title	<b>Introductory Quantum Science for engineers</b>	Course Code	PHY508			
Dept./ Specialization	SH (Physics)	Structure (LTPC)	3	1	0	4
To be offered for	UG/PG	Status	Core <input type="checkbox"/>		Elective <input checked="" type="checkbox"/>	
Faculty Proposing the course	Tapas Sil	Type	New <input checked="" type="checkbox"/>		Modification <input type="checkbox"/>	
Recommendation from the DAC		Date of DAC				
External Expert(s)						
Pre-requisite		Submitted for approval				
Learning Objectives	<ul style="list-style-type: none"> <li>To develop in the student, an awareness of situations in engineering, which need ideas of quantum mechanics.</li> <li>The course emphasizes conceptual understanding essential mathematics is for understanding and using quantum mechanics.</li> <li>To enable the student with those aspects of quantum mechanics, which are necessary to begin to deal with microscopic systems.</li> </ul>					
Learning Outcomes	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>Understand the fundamental concepts and quantum mechanical processes in the nature.</li> <li>Apply principles of quantum mechanics to calculate observables on known wave functions or potentials.</li> <li>Pursue more advanced courses such as quantum communications, quantum computation, quantum optics, nanophotonic devices etc.</li> </ul>					
Contents of the course <i>(With approximate break-up of hours for L/T/P)</i>	<p><b>Introduction:</b> The bizarre aspects and continuing evolution of quantum mechanics, and how we need it for engineering modern technology. Blackbody radiation, The photo-electric effect, Atomic spectra, The Frank-Hertz experiment, Compton effect, Wave-Particle duality, Wavefunctions, Expectation values, Uncertainty principle. [L12+T3]</p> <p><b>Schrodinger's wave equation:</b> Getting to Schrodinger's wave equation. Solution of stationary-state Schrodinger equation for one dimensional bound state problems. Potential barrier and tunneling and applications such as, Esaki diode, scanning tunneling microscope, etc.; Particle in 3D box and related examples (quantum dot, quantum wire etc); Quantum mechanical measurements and wavefunction collapse [L12+T3]</p> <p><b>Aspects of angular momentum and spin:</b> Angular momentum operators. Stern-Gerlach experiment—spin. Solution of hydrogen atom problem. [L10+T4]</p> <p><b>Introduction to Quantum information :</b> Quantum cryptography, Entanglement, Quantum computing, EPR paradox, Bells inequality [L8+T2]</p>					
Text Book	<ol style="list-style-type: none"> <li>Devid J. Griffiths and Darrell F.Schroeter," Introduction to quantum mechanics", (Cambridge University Press, 3rd edition, 2019)</li> <li>P A M Dirac, "The Principles of Quantum Mechanics", (WWW.Snowballpublishing.com, 2013)</li> </ol>					
Reference Books	<ol style="list-style-type: none"> <li>Asher Peres, ``Quantum Theory: Concepts and Methods'', (Kluwer Academic Publishers, 1993)</li> <li>D. A. B. Miller, "Quantum Mechanics for Scientists and Engineers," (Cambridge University Press, 2008)"</li> </ol>					